Assignment

Chapter 01 – Chemical Reactions and Equations

Q1. Which of the statements about the reaction below are incorrect?

$$2PbO(s) + C(s) \rightarrow 2Pb(s) + CO_2(g)$$

- a) Lead is getting reduced
- b) Carbon dioxide is getting oxidized
- c) Carbon is getting oxidized
- d) Lead oxide is getting reduced

i. (a) and (b)

ii. (a) and (c)

iii. (a),(b) and (c)

iv. All

Q2. Fe2O3 + Al \rightarrow Al2O3 + 2Fe

The above reaction is an example of a

- a) Combination reaction
- b) Double displacement reaction
- c) Decomposition reaction
- d) Displacement reaction

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