

Assignment

Chapter 02 – Acids, Bases and Salts

2.1.3 How do Metal Carbonates and Metal Hydrogencarbonates react with Acids?

Q. Metal compound A reacts with dilute hydrochloric acid to produce effervescence.

The gas evolved extinguishes a burning candle. Write a balanced chemical equation for the reaction if one of the compounds formed is calcium chloride.

Email to: techtutor.megh@gmail.com